

SOLUTION MINING RESEARCH INSTITUTE

**812 MURIEL STREET
WOODSTOCK, ILLINOIS 60098
815-338-8579**

**MEETING
PAPER**



PRESENTED AT SMRI MEETING
AUSTIN, TEXAS
MONDAY, APRIL 23, 1990

IMPURITIES IN RAW BRINE FROM
BEDDED SALT DEPOSITS

BY
DONALD R. RICHNER

PB-KBB INC.
SUBSURFACE SYSTEMS AND TECHNOLOGY
11767 KATY FREEWAY, SUITE 810
HOUSTON, TEXAS 77224
(713) 496-5590

1. INTRODUCTION

There are numerous gaseous and water-soluble solid substances contained within a bedded evaporite deposit that can affect the operation of a brine field, pipelines, or the chemical plant. Gaseous substances include free ammonia (NH₃), hydrogen sulfide (H₂S) methane (CH₄), carbon dioxide (CO₂) and carbon monoxide (CO). Soluble solid compounds contain calcium, magnesium, potassium, and iron, as well as sulfates, sulfides, carbonates and silicates. Liquid hydrocarbons are also occasionally encountered.

Most of these impurities are contained within the interbedded mudstones, marlstones and shales found within the evaporite deposit. It is important, therefore, that the entire evaporite sequence to be solution-mined within a New Brine Field be continuously cored. Cores should be geologically described, properly sampled, and chemically analyzed for potential injurious substances as promptly as possible after retrieved from the core hole. A lapse of a few weeks before chemical analysis will permit significant degassing of the core.

©2023 - Solution Mining Research Institute
Full Paper is Available in the SMRI
Library(www.solutionmining.org)